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Effect of Curing Profile on Fly Ash Geopolymer with Slag as Supplementary

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Abstract

The aim of this research is to evaluate the curing temperature at which the fly ash based geopolymer can achieve its best mechanical properties with the presence of supplementary calcium source. Wide range of curing temperature from 55°C-85°C was imposed to sample GP (without any calcium compound) and sample GB (with 15% of slag as supplementary material) to have their comparative performance. The prime objective is to find out the right curing temperature in a manner to obtain the optimized strength under compression.

Keywords: Slag, Geopolymer, Fly ash, Curing, Activator.

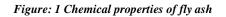
Introduction

In earlier study, the aim was inclined towards the effect of curing temperature as well as curing duration of fly ash and kaolinite based geopolymers which proves that higher temperature for more than a couple of hours seemed better for the development of compressive strength [1]. Proper parametric study on blended geopolymer has not yet done. Though literature review gives some data regarding optimal value of different mixing parameter. Before any further research or extension a detail understanding on the formation of fly ash geopolymer, it's control parameter and also the role of different supplementary material are needed. In last decade, a positive research outcome is low calcium fly ash based geopolymer cement and concrete [2-5]. Its preparation includes activation in an alkaline medium and curing at moderate temperatures. It is the fact, that Low-calcium ash is more as a source material of geopolymer rather than that containing high calcium. High calcium in fly ash as base material may disturb the process polymerization or change the microstructure [6]. Research has been done on slag a supplementary material in fly ash based geopolymer to have a favorable effect.[7]. Addition of calcium compound to a fly ash system has been proved that it brings quick setting behavior and enhance strength [8].In chemical terminology, it shows that calcium compound subjected to alkaline medium forms geopolymery and C-S-H component [9,10]. As we know, excessive presence of calcium creates highly unstable calcium hydroxide which is extremely prone to carbonation [10]. Mechanism of geopolymers involves the polycondensation geopolymeric reaction of precursors i.e. aluminosilicate oxide with alkali polysiliates polymeric Si-O-Al vielding bond [11,12,13,14]. The basic polymeric formula can be expressed as $M_n[-(Si - O_2)_z - Al - O]_n$.wH₂O where M is the alkaline element, z is 1,2, or 3 and n is the degree of polycondensation [13]. Now any alkali cation can be used as (M) in the polymeric reaction but most of the study has entertained on sodium or potassium [7, 9-11]. The incorporation of calcium is much important to consider as an alkali cation which may act as a charge compensator of aluminium to form an amorphous geopolymeric gel rather than any part of a basic geopolymeric structure. It is already found that in higher value of pH the alkali activation of metakaoline in presence of calcium hydroxide form complete amorphous sodium alumino silicate which is similer to that when metakaoline was activated in absence of Ca(OH)₂[12, 13]. In this type of geopolymer, C-S-H can be treated as the secondary product. The present investigation is targeted towards the findings of the optimal curing profile to have the best mechanical properties like strength.

Experimental

Materials

Class F fly ash used in the research was collected from Kolaghat Thermal Power Plant near Kolkata, India. About 75% of particles were finer than 45 micron and Blaine's specific surface was $380m^2/kg$. The chemical composition of fly ash is given in Figure 1. The blast furnace slag used was in powdered form having specific gravity 2.8, bulk density 1236 kg/m³, consisting of 39.07% CaO. The average particle size of blast furnace slag was varied between 35μ to 65μ . The chemical composition of blast furnace slag is given in Figure



2.

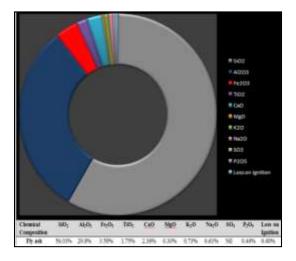
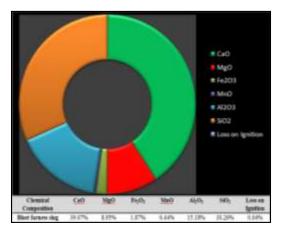


Figure: 2 Chemical analysis report of blast furness slag



Laboratory grade sodium hydroxide in pellet form (98% purity) and sodium silicate solution (Na₂O= 8%, SiO₂ =26.5% and 65.5% water) with silicate modulus ~ 3.3 and a bulk density of 1410 kg/m³

was supplied by Loba Chemie Ltd, India. Scanning electron micrographs of fly ash and Blast furness slag is given in Figure 3 and Figure 4.

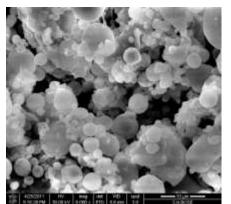


Figure: 3 SEM of Fly ash at 6000x zoom

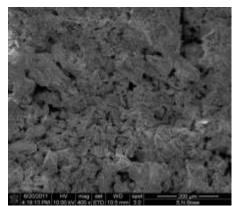


Figure: 4 SEM of Blast furness slag at 400x zoom

Preparation of Solution, Specimens and Testing

The alkaline activating solution was prepared by dissolving required quantity of sodium hydroxide pellets directly into water .The activator solution (sodium hydroxide and water) was left at room temperature for 24 hours after that predetermined quantity of sodium silicate solution was added 3 hours before casting of geopolymer specimens.

In a Hobart mixer, fly ash, with or without blast furnace slag (according to Table 1) was mixed with predetermined quantity of activator solution

Table 1(a). Dry mix combination

Sample Id→	GP	GB
Base Material (Bm)→	Fly ash	Fly ash
Supplementary Material (Sm)→	NIL	GGBS (15% of base Bm plus Sm)

Sample ID	Activator						
	%Na ₂ O	Silicate Modulus					
GP	6%	0.5					
		1					
		1.5					
GP	8%	0.5					
		1					
		1.5					
GB	6%	0.5					
		1					
		1.5					
GB	8%	0.5					
		1					
		1.5					

Table 1(b). Activator Combination

Table 1(c).	Hot	Curing	Exposure	regimes
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Curing Duration →		24 H	Iours		48 Hours				
Curing Temperature →	55°C	65°C	75°C	85°C	55°C	65°C	75°C	85°C	

Moto of the present work

The motto of the work is to find out the impact of different curing period on the mechanical property like compressive strength for slag-blended fly ash based geopolymer composite. To determine the optimal curing period the other parameters influencing geopolymerisation were kept controlled.

Result and discussion

Compressive Strength

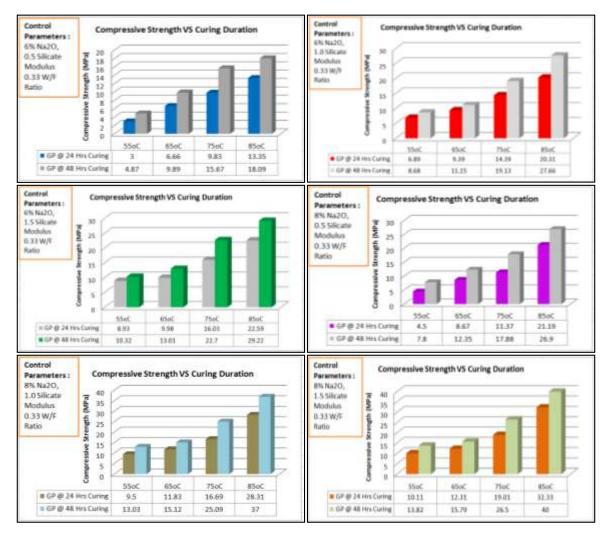
The compressive strength of the cube specimens were evaluated according to ASTMC109 by using the digital compressive strength testing machine. Every specimen under different curing environment were taken out of oven and put in room temperature for 1 day. The compressive strength was carried out to evaluate the strength development for the samples. The samples were subjected to compression at 3 days. Fly ash based geopolymer needs sufficient time for the geopolymerization process to happen, and therefore, in order to increase the dissolution of reactive species sufficient amount of heat influx is needed for strength gain. Here the term heat influx indicates the product of curing temperature and curing time. The graphical demonstration depicts the successive increments in compressive strength with prolong curing. The incorporation of supplementary slag in fly ash based geopolymer upto 15% of fly ash brings better chemistry in geopolymerization. Here calcium plays better role

in geopolymerization than sodium.

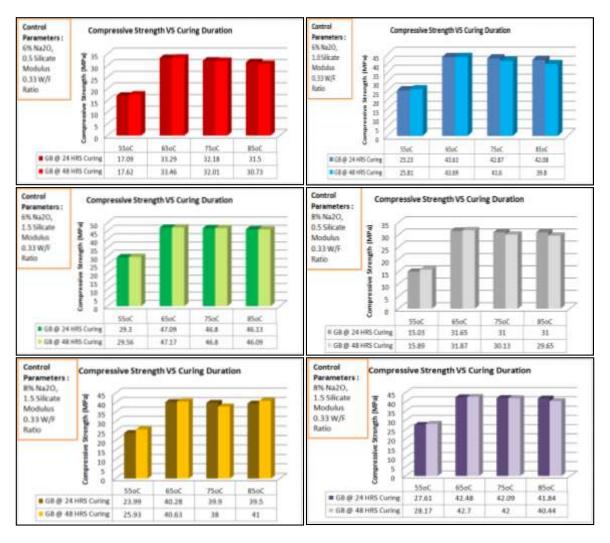
Sample	Activator			Compressive Strength (MPa) after 3 days from														
ID	Silicate Modulus	%Na ₂ O		55°C	65°C	75°C	85°C		55°C	65°C	75°C	85°C						
	0.5			3.0	6.66	9.83	13.35		4.87	9.89	15.67	18.09						
GP	1		8	6.89	9.39	14.39	20.31	8	8.68	11.15	19.13	27.66						
	1.5		-	8.93	9.98	16.03	22.59	ays	10.32	13.01	22.7	29.22						
	0.5	8%	Day	4.5	8.67	11.37	21.19	Da	7.80	12.35	17.88	26.90						
GP	1		8%	8%	8%	8%	8%	8% 7	8% 7	9.5	11.83	16.69	28.31	7	13.03	15.12	25.09	37.00
	1.5			for	10.11	12.31	19.01	32.33	for	13.82	15.79	26.50	40.00					
	0.5		uring	17.09	33.29	32.18	31.50	gu	17.62	33.46	32.01	30.73						
GB	1	6%	ini	25.23	43.61	42.87	42.08	uring	25.81	43.69	41.60	39.80						
	1.5	Ū	C	29.30	47.09	46.80	46.13	ū	29.56	47.17	46.80	46.09						
	0.5					15.03	31.65	31.00	31.00		15.89	31.87	30.13	29.65				
GB	1	8%		23.99	40.28	39.90	39.50		25.93	40.63	38.00	41.00						
	1.5			27.61	42.48	42.09	41.84		28.17	42.70	42.00	40.44						

Table 2. Compressive Strength of typical specimens

Figure: 5 Elementary exhibition of Table 2 in graphical format to evaluate the effect of Curing Duration on the compressive strength of geopolymer under different control features



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Actually, Ca⁺⁺ acts as charge balancer of aluminum which emphasis the faster formation of amorphous geopolymeric structure. It may be assume that optimal presence of calcium increases the rate of polycondensation by influencing the resolution of reactive species. It could be observed that GP shows corresponding sample increment in compressive strength for longer curing period, whereas blended geopolymer specimens does not influence in that way. It is also noticeable that few of the GB sample exhibits little drops in strengths due to longer curing duration. It is because of excessive pressure on the hardened geopolymer structure caused by the water trying to be extruded. It can be concluded from the compressive strength data that one day curing duration is sufficient for fly ash geopolymer blended with 15% slag of fly ash.

Conclusion

1. Specimen GB shows very small percentage of increment in compressive strength for longer curing duration for a curing temperature upto 65°C, but every case this increment is not greater than 1%

even. Furthermore a curing duration of 48 hours for a curing temperature greater than 65°C gives poor result in connection in compressive strength.

2. Non-blended geopolymer GP undergoes to sufficient increments in compressive strength with prolong curing duration. Maximum compressive strength is obtained by a typical GB specimens subjected to 65° C curing temperature for 24 hours. While the highest value of compressive strength for GP specimens is 40 MPa for a curing profile of 85° C at 48 hours.

3. The geopolymerization chemistry is much effected by the introduction of 15% GGBS with fly ash. This supplementation brings optimal strength in shorter time period of curing. It is because of the higher rate of polycondensation in the presence of Ca^{++} cation as charge balancer of aluminiuam.

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